

EXAMINATIONS COUNCIL OF ESWATINI Junior Certificate Examination

 Science
 414/02

 PAPER 2
 2019

MARK SCHEME

{414/02}

Confidential

1	(a)	metre rule; accept tape measure	[1]	
	(b)	35 – 10;		
		25 seconds;	[2]	
	(c)	0.4 m; seen any where		
		S = d/t / 0.4/25;		
		= 0.016;	[3]	
2	(a)	heat solution + to evaporate;		
		condense;		
		water collected as distillate;		
		salt remains as residue;		
		distillation; seen anywhere	[max 3]	
		OR		
		labelled diagram of distillation apparatus earns full marks		
		all apparatus included;		
		apparatus labelled i.e. Bunsen burner, (distillation) flask, condenser, vessel;	correct collecting	
		residue /salt + distillate/ water;		
	(b)	can be separated by physical methods/ filtration/ distillation/ crystallis	sation/ evaporation/	
		separating funnel/ magnetism/ chromatography;		
		composition is not fixed/ variable/ aw;		
		reversible;		
		no new substance formed;	_	
		energy not involved/ taken in/ absorbed / released; any two,	[max. 2]	
3	(a)	shiny/ lustrous;		
		malleable;		
		sonorous;		
		conduct heat/ electricity;		
		ductile/;	. 01	
		high melting and boiling point; any two	[max. 2]	
	(b)	F;	[1]	
	(c)	is insoluble in water;		
		less dense than water;	[2]	

4	(a)(i)	palisade cell;		[1]
	(ii)	nucleus;		[1]
	(iii)	site for photosynthesis;	[1]	
	(b)	contains a group of tissues/xylem and phloem/ epidermis;		
		that perform a certain function/ photosynthesis;		[2]
	(c) broad;			
		network of veins;		[2]
5	(a)	heated particles expand;		
		and become lighter;		
		lighter particles rise and cold denser particles go down;		[3]
	(b)(i) vacuum (ii) no particles;			[1]
		conduction and convection involves particles;	[2]	
	(c)	electrons from the cloth;		
		are transferred to the ruler;		[2]

6 (a) Table 6.1

steps	description	explanation		
Step 1	stirring/	to mix reactants and evenly distributed heat/		
	heating	to make reaction faster;		
step 2	filtration	to remove <u>excess</u> copper(II) oxide;		

[4]

salt	starting materials		
calcium sulfate	calcium	dilute sulfuric acid	
potassium chloride	potassium hydroxide	hydrochloric acid	

							[3]
-	(-\(!\		d.,				
7		sperm			_		[1]
	(ii)	label li	ne to the testes labelled	d F ; A	F on test	es	[1]
	(b)	testost	erone;				[1]
	(c)	Tubero	culosis/hepatitis B/cervi	cal cancer/mening	gitis/pneum	nonia [max. 2]
8	(a)	F r	epulsion;				
		G a	attraction;				[2]
	(b)	a	at least 2 field lines fron	n each pole;			
		C	correct direction of field	lines;			[2]
	(c) magnet gets struck/hammered/AW;						
	magnetic domains disorganised;						[2]
9	(a)	particles are close together;					[1]
(b) diffusion;							
		particles move from where they are more concentrated (in the kitch					
		to whe	re they are less concer	ntrated (Thabo):	•	·	[3]

[1] 10 (a) erector muscle; (b) collect water (+ salts) from blood capillaries; water moves up sweat ducts into sweat pores; forms layer of water on skin; evaporates using energy from the skin; any three [max. 3] 11 total resistance = 1.5 + 2= 3.5Ω: I = V/RI = 1.5/3.5; = 0.43 A;[3] **SECTION B** 12 (a) (i) $v = 35 \text{ (cm}^3)$; M = 75; [2] (ii) put measuring cylinder on a flat surface; eye must be level with bottom of meniscus; [max. 1] (b) tie stone with string (of negligible volume); (gentle) lower the stone into the water until it is fully immersed and record volume (as V2); volume of stone = $V_2 - V_1$; measure mass of the stone using a triple beam balance/ electronic balance: a named balance: [max. 4] divide mass by volume to get the density; (c) (i) suitable scale covering more than half the grid; all points correctly plotted;; 5 points earn 1 mark points joined + smooth curve; [4] (ii) check from the pupils graph at 7cm; [1] (iii) the closer the plant from the lamp the more bubbles are produced; [1] (iv) test: introduce a glowing splint into the test tube; result: glowing splint relights/ rekindles; [2] (v) to improve accuracy/ to reduce (experimental) errors; [1]
 (vi) get a plant with variegated leaves;
 destarch the plant;
 expose the plant to broad daylight for few hours;
 keep temperature/ light intensity/ carbon dioxide

concentration constant;
remove the leaves from the plant;
use iodine solution to test for the presence of starch;

the parts of the leaf that have chlorophyll/ green part turns blue-black; [max. 4]

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